

## **2.5.4 (1)**

**Retest for improvement**

SANTHIRAM MEDICAL COLLEGE, NANDYAL (A.P)  
Improvement EXAMINATION – BIOCHEMISTRY PAPER –II

**II. THEORY QUESTION – 80M**

**ESSAY QUESTION**

**2X15=30**

1. Describe in detail the process of transcription in Prokaryotes. Add a note on post transcriptional modifications and inhibitors of transcription. (8+4+3= 15 Marks)
2. Explain the role of plasma buffers and respiratory mechanism in maintenance of acid base balance in the body. Add a note on Metabolic Acidosis. (5+5+5= 15 Marks)

**Write short notes on:**

**10X5=50**

3. A 15yr old girl presented to OPD with abdominal distension and Jaundice. On examination bilateral Kayser-Fleischer ring noted in both eyes. LFT-Normal.

1. What is the likely diagnosis? (1 M)
2. What is the biochemical basis? (2M)
3. What biochemical investigations has to done ? (1M)
4. What is the treatment ? (1M)

4. Mechanism of action of Group I hormones.

5. Structure & Metabolic pathways in Mitochondria.

6. Immune responses.

7. Proto oncogenes.

8. Structure of collagen.

9. Acute phase proteins.

10. Steps of Ureacycle and regulation.

11. a) Derivatives of glycine

- b) Any 4 SAM involved reactions

12. A 55yr old man complained severe pain in the joints. He is a non-vegetarian and consumed alcohol occasionally. His laboratory findings are as follows.

1. S.Uricacid -12mg/dl    2. Bl.urea -25 mg/dl    3. Urinary uric acid -2.5g/day    4. Urine PH -4.5

1. Write the diagnosis. (1M)

2. Which metabolic pathway is affected ? (1M)

3. Describe the types and treatment of the disease. (2M)

4. What are the normal reference ranges for the above parameters.(1M)



# SANTHIRAM MEDICAL COLLEGE NANDYAL



1<sup>st</sup>

Year BSc

I/II/III Internal Assessment Examination

I II III

Subject Biochemistry Paper I/II

Roll No. 21M102017021

No. of Additional Sheets 1

Dr. B. Indumathi  
Signature of Invigilator

B. Dheeraj Rddy  
Signature of Student

(To be filled by the Examiner)

Max Marks :

Marks Question Wise

Q.No.	Page No.	Marks
1		9.5
2		9.5
3		3.5
4		3
5		2
6		3.5
7		3
8		3
9		3.5
10		2.5
11		2
12		3
13		
14		
15		

$$\frac{48}{80} + \frac{10}{20} = \frac{58}{100}$$

Signature of Examiner

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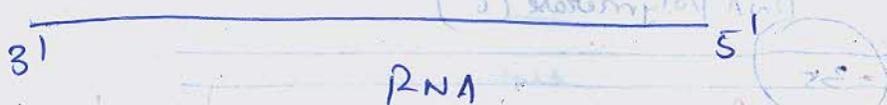
## ⇒ ESSAY'S

### 1) Transcription in prokaryotes →

\* The Synthesis of RNA from the DNA is known as transcription



↓ (Transcription)



\* Transcription in prokaryotes occur in 3 process and they are:

i) Initiation

ii) Elongation

iii) Termination.

\* Transcription depends upon the following

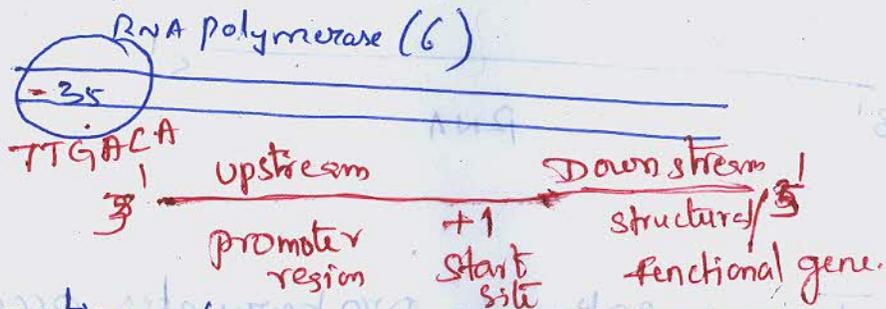
4 factors → template strand, (coding strand etc (3-5' strand))

## i) Initiation →

write few lines about RNA polymerase enzyme

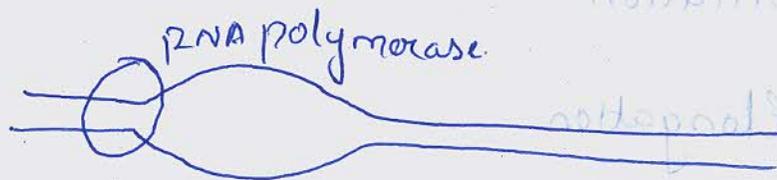
### \* Formation of closed Complex

- Attachment of RNA polymerase to the promoter region
- Scanning of DNA will done by the sigma factor ( $\sigma$ )



### \* Formation of open complex

- local binding of DNA at the region of TATA - box



### \* Initiation of RNA Synthesis

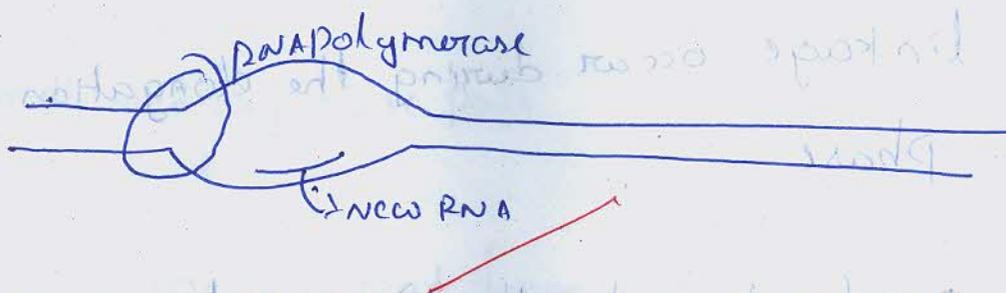
- Synthesis of RNA from the template strand

- Complementary to RNA polymerase by the template strand

- During this phase there is no need of the primer

- Short stretch of RNA get synthesized

(5-10 RNA)



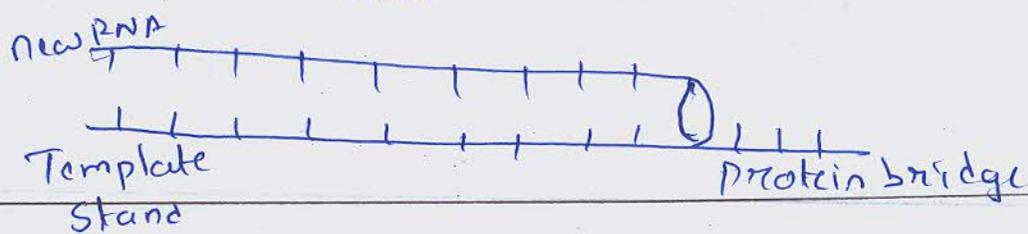
\* Promoter clearance

- The RNA get shifted to the transcription site and get moved away from the promoter region



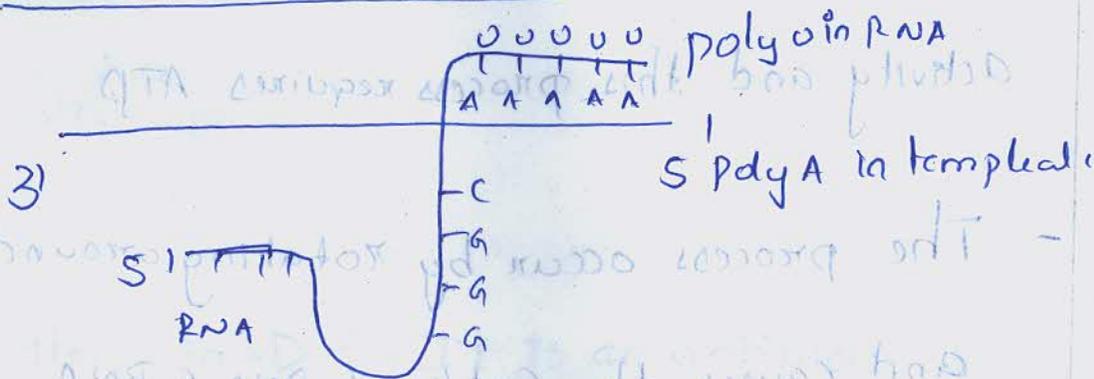
## ii) Elongation →

- \* At this phase, the ribonucleotides are added to the chain and the chain increases continuously
- \* Formation of 3'-5' phosphodiester linkage occur during the elongation phase
- \* Proofreading by the RNA polymerase does not occur
- \* Protein bridge provide the nucleotide to the template strand for the RNA Synthesis
- \* Nucleotides are added to 3' end of molecule



iii) Termination  $\Rightarrow$   
Rho independent termination

\* Intrinsic termination



Complementary base pair

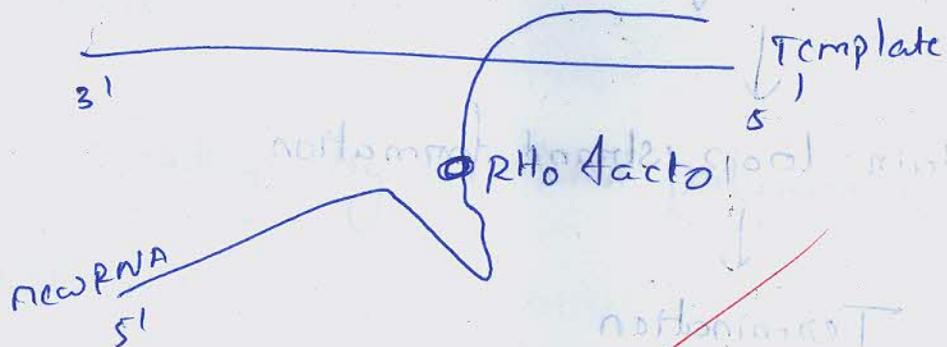
↓  
Hair loop strand formation  
↓  
Termination

\* The occurrence of poly U in the RNA is due to rich signals cause the termination in the transcription

\* Rho protein dependent transcription

← The activated RNA activate the Helicase activity and this process requires ATP

- The process occur by rotating around and causes the cut's of DNA & RNA



## ⇒ Inhibitors

i) Acitomyacin-D

ii) Rifampicin

\* Acitomyacin-D → It is an antibiotic and

It intercalated into Double  
Strand DNA & blocks the  
RNA polymerase

13  
\* Rifampicin → It is antitubercular drug

It bind with  $\beta$ -subunit of  
RNA polymerase and prevents  
the transcription

⇒ Post-transcriptional modifications

- Post transcriptional modifications occurs  
in 2 categories →

i) Proteolytic cleavage

ii) Covalent modifications

92

## 2) Acid-base balance (buffers) $\rightleftharpoons$

\* The body has of three lines of defence mechanisms for the regulation of blood pH & acid base balance.

\* The mechanisms are  $\rightleftharpoons$

- i) Blood buffer (plasma)
- ii) Renal mechanism
- iii) Respiratory mechanism

### i) Blood buffer $\rightarrow$

\* The solution of the weak acid and the strong base is known as buffer

\* Blood buffers are of 3 types

- Protein buffer
- ~~Phosphate buffer~~
- Bicarbonate buffer

## Bicarbonate buffer

\* It is most predominant buffer system of the extra cellular fluid

\* It depend on the sodium bicarbonate and carbonic acid



\* It's ratio is 20:1

\* It's pka value is 6.1

## Phosphate buffer

It depend on the dihydrogen phosphate and dihydrogen sodium phosphate

It is defined as intracellular buffer due to low conc of plasma

It's ratio is 4:1

It's pka value is around 6.8

## - Protein buffer

\* Plasma protein and the hemoglobin both together constitute the protein buffer

- Its  $pK_a$  value is 6.7

- It is mainly due to presence of imidazole group.

## ii) Respiratory mechanism →

\* The Respiratory mechanism provide a rapid action mechanism to the acid base balance

\* It is activated by regulating of the carbonic acid



\* As this mechanism has rapid action, but it for the short regulatory period.

⇒ lungs

i) Alkalosis →

- Decrease in the conc of  $PCO_2$

- Increase in conc of  $H_2CO_3$  &  $PCO_2$

- It causes hypoventilation

~~ii) Acidosis →~~

- Decrease in the conc of

$H_2CO_3$  &  $PCO_2$

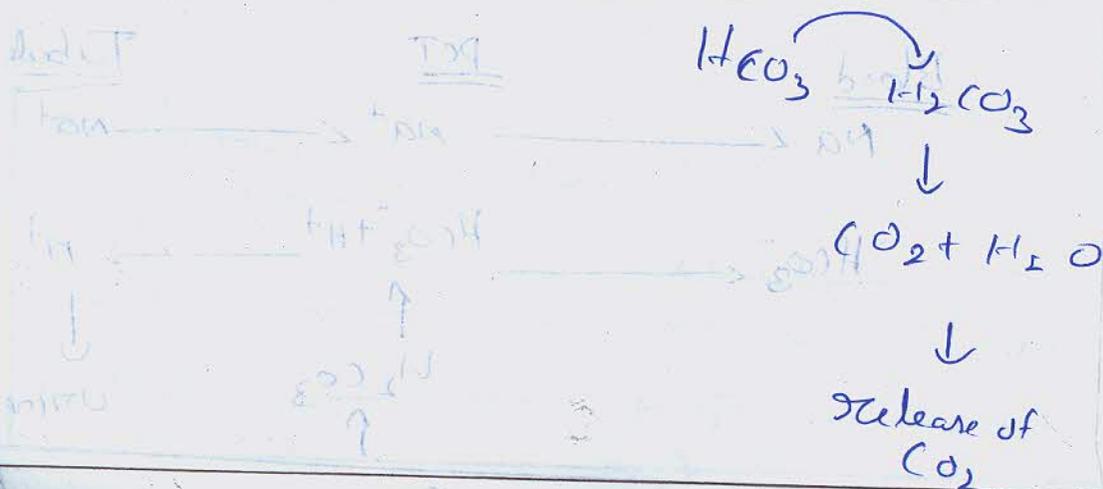
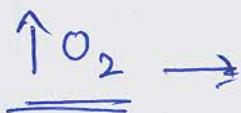
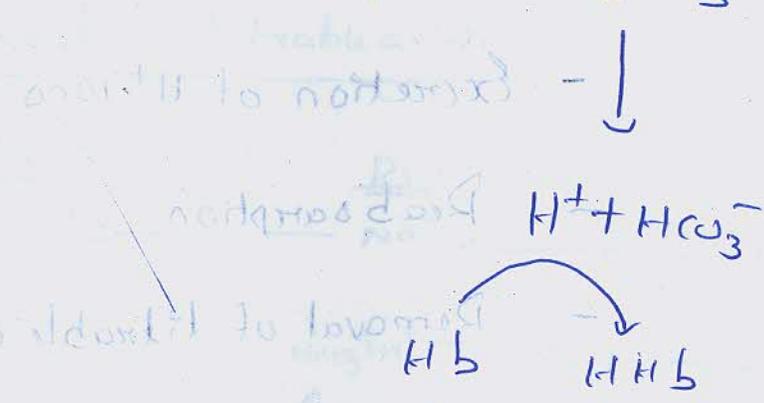
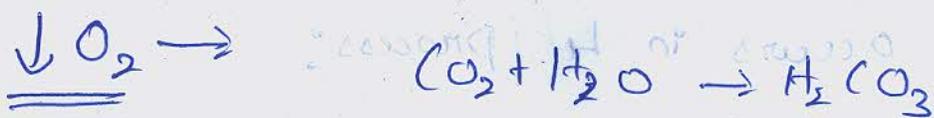
- Increase in conc of  $PCO_2$

→ Hemoglobin

← Respiratory

→ It binds to H<sup>+</sup> ions and helps in the transport of CO<sub>2</sub> & HCO<sub>3</sub><sup>-</sup>

→ Tissues causes decrease in O<sub>2</sub> and increase in CO<sub>2</sub>



dy

er

### iii) Renal mechanism $\Rightarrow$

\* Renal mechanism plays a major role in acid base balance

\* As this mechanism provide a permanent solution to the acid-base disorders

\* It occurs in 4 process:

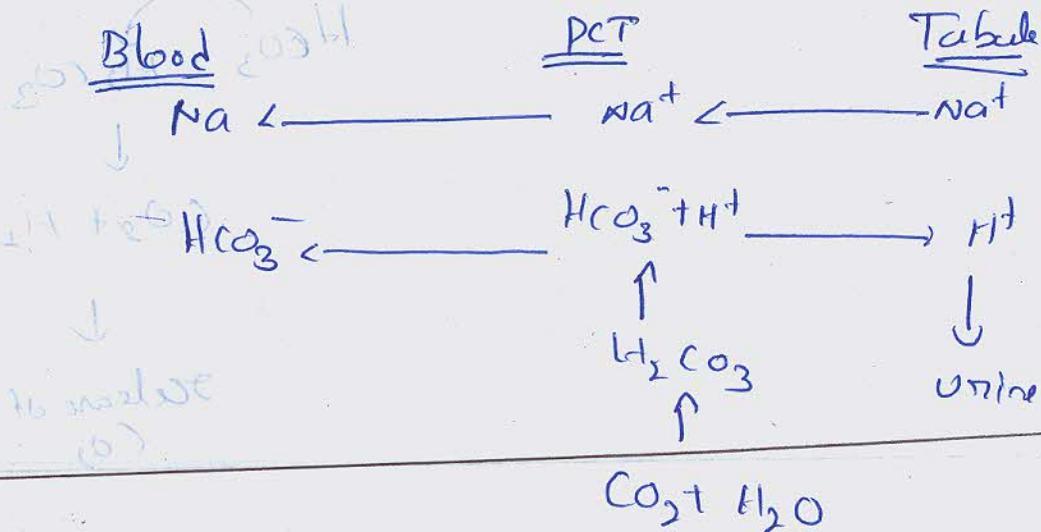
↓ - Excretion of  $H^+$  ions

↑ - Reabsorption

- Removal of titrable acid

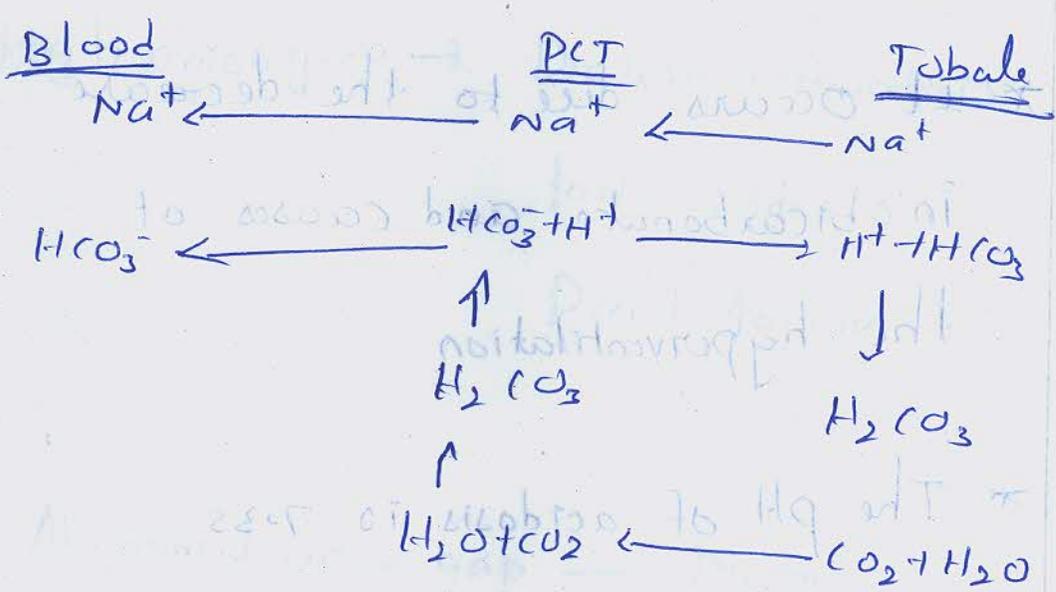
- Formation of  $NH_4^+$

- Excretion of  $H^+$  ions

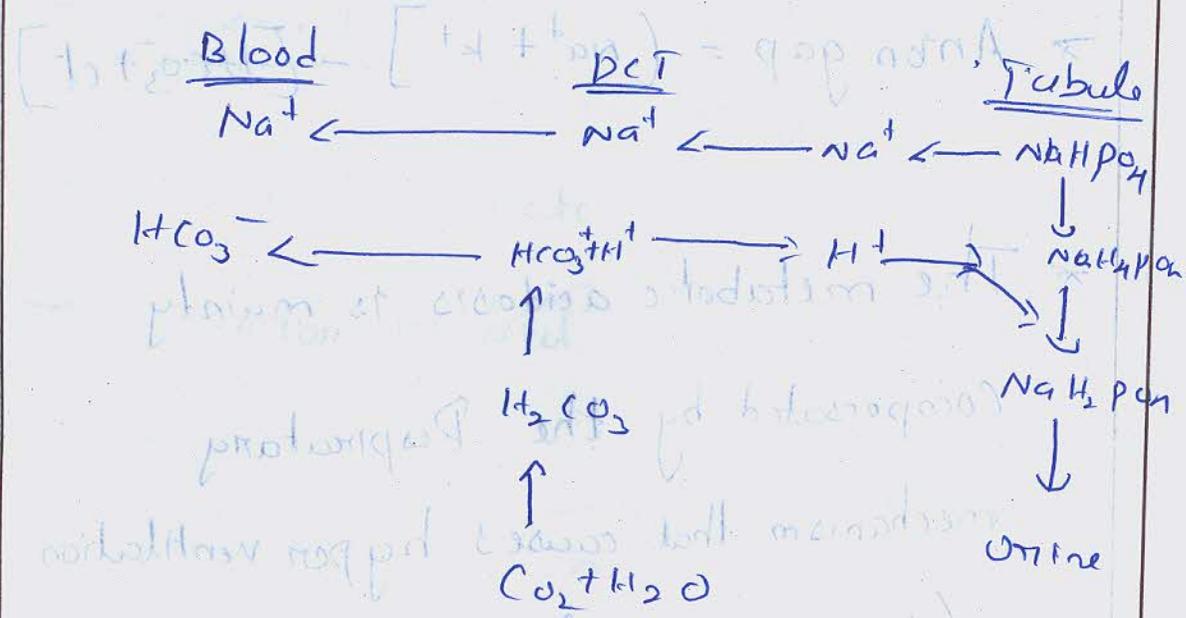


Reabsorption

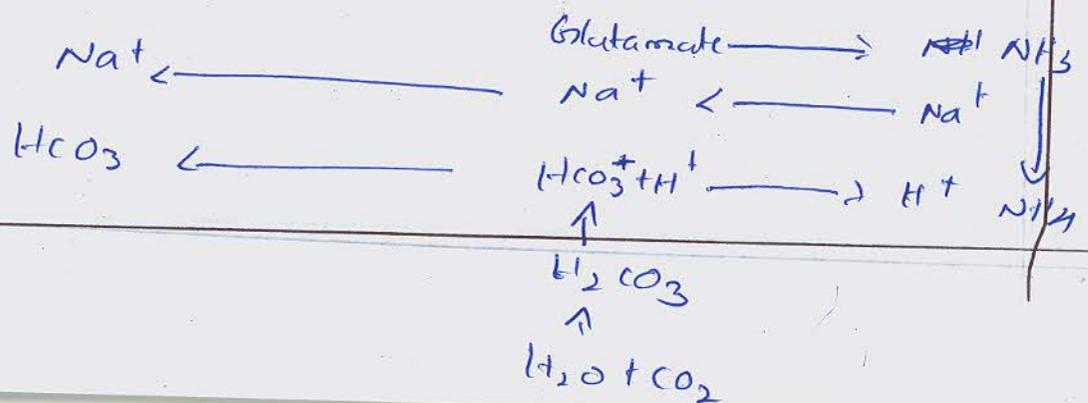
Metabolic Acidosis



Removal of titratable acids



Formation of NH<sub>4</sub>



## ⇒ Metabolic Acidosis

\* It occurs due to the decrease in bicarbonate and causes of the hyperventilation

\* The pH of acidosis is 7.35

\*  $\text{HCO}_3^- < 22 \text{ mol/l}$

\* Anion gap =  $[\text{Na}^+ + \text{K}^+] - [\text{HCO}_3^- + \text{Cl}^-]$

\* The metabolic acidosis is mainly compensated by the Respiratory mechanism that causes hyperventilation ( $\downarrow \text{pCO}_2, \downarrow \text{H}_2\text{O}$ )

## ⇒ Causes

- High anion gap → Diabetic ketoacidosis

Lactic acidosis

Renal failure

- Abnormal anion gap → Diarrhea

Renal failure

3

## → Treatment

→ Sodium bicarbonate

→ ~~monitor kt level~~

93

⇒ Short's:

3)

i) The patient is suffering from the Wilson's disease



ii) Biochemical basis:

- Hepato lenticular degeneration
- Copper binding ATPase mutation

iii) Biochemical investigations:

- Accumulation of copper in liver
- Damages lenticular nucleus of brain

→ 2 marks

### iv) Treatments

\* Less intake of copper in diet



Wilson's disease

(i) Biochemical basis:

- Defective lysosomal degradation
- Copper binding ATPase mutation

(ii) Biochemical investigations:

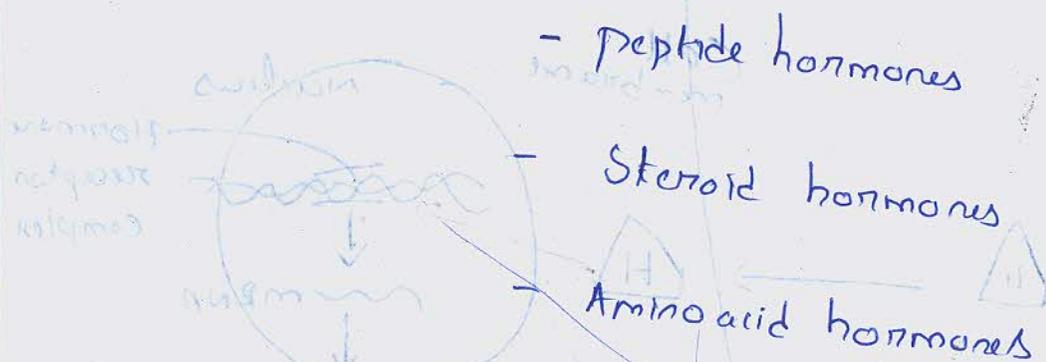
- Accumulation of copper in liver
- Damaged lysosomes
- ...

## 4) Group-1 hormones →

### \* Hormones →

- It is defined as the organic substances that produced in small amounts by specific tissue
- Hormones are commonly regarded as chemical messengers
- Hormones are classified into 2 categories:

### i) Chemical messengers →



### ii) mechanism of action →

- Group-1
- Group-2

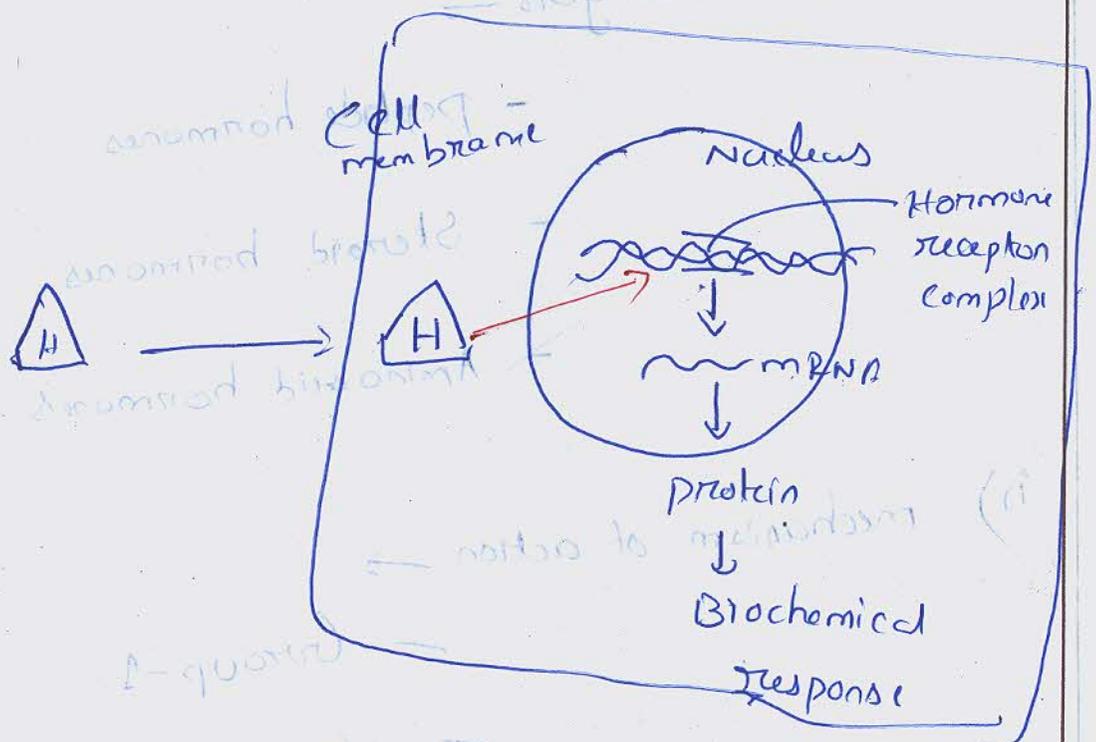
## ⇒ Group - 1 hormones

- \* These group of hormones bind with the intracellular receptors for the formation of hormonal complex
- \* These are lipophilic in nature and

the derivatives of cholesterol

Eg: Estrogen, calcitriol, FSH, LH,  
steroid hormones, progesterone, estrogen.

## \* Mechanism



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When the Hormone (H) attaches to the receptor  
then the negative effector gets released

↳ positive effector gets attached to the  
DNA

↳ transcription, translation of protein will  
occur

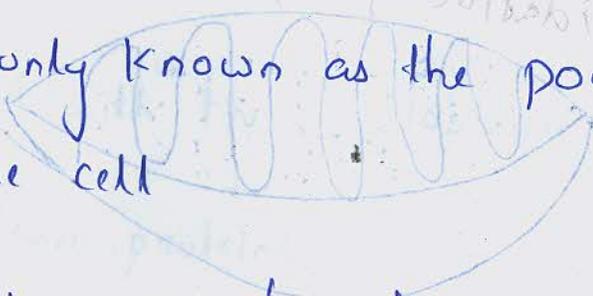
3'

ddy  
t

## 5) Mitochondria →

\* mitochondria is one of the important cell organelle

\* It is commonly known as the power house of the cell



\* It provide the ATP for the energy to the cell.

\* The mitochondria consist of cristae, matrix, inner & outer mitochondrial membrane.

\* The mitochondria play a major role (site) into the metabolic pathways.

⇒ metabolic pathways

- In the electron transport chain the inner mitochondrial membrane act as site for synthesis

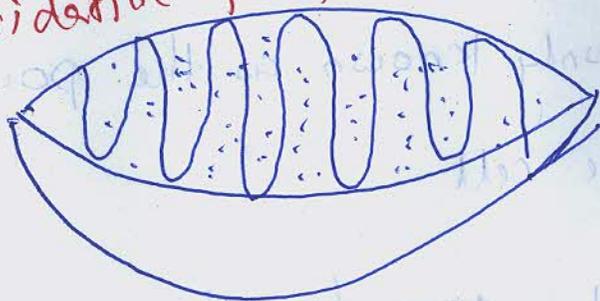
- Heme synthesis

- Glucose metabolic pathways

- TCA cycle

-  $\beta$ -oxidation  
oxidative phosphorylation

2



← Mitochondria (2)

It provides the ATP for the energy

to the cell

The mitochondria control of cellular

homeostasis, energy, and

metabolism

The mitochondria play a major role in

the metabolic pathways

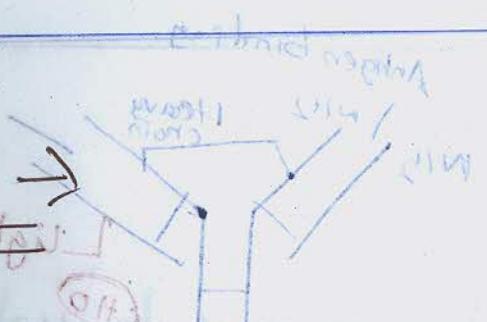
⇒ metabolic pathways

- In the electron transport chain the

protons are pumped across the

membrane

## 6) Immune response →



- Immunoglobulin is a special group of proteins

- It is associated with the  $\gamma$ -globulin

Fraction of plasma proteins

- It is a functional form

### ⇒ Structure

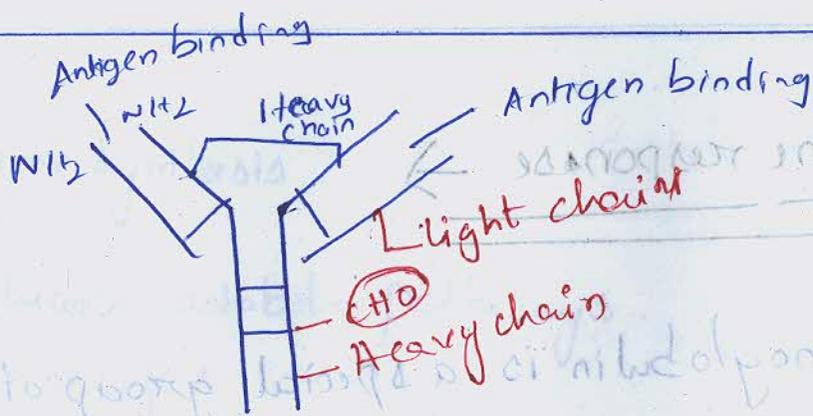
- Immunoglobulin consist of 2 heavy chains and 2 light chains

- The molecular weight of heavy chain is 50000 to 75000 and the light chain is around 25000

- It is a  $\gamma$ -shaped tetramer

- 450 amino acids present in heavy chain

- 200 amino acid in light chain



## ⇒ Classification

- Immunoglobulins are classified into 5

Classes:

Ig G

Ig A

Ig M

Ig D

Ig E

- It consists of 2 types of light chains

\* kappa

\* lambda

\* IgG :

\* Mpl \*

It is most abundant class of immunoglobulin

It composed of single Y-shaped unit

It can traverse blood vessels

It is only the immunoglobulin that cross the

placenta & transfer mother immunity to newborn  
& protect against the infections.

\* IgA :

It occur as single (or) dimer unit

It held together by J chain

It is found in body secretions such as sweat,  
tears etc

It is most predominant antibody in the  
colostrum

## \* IgM:

- It is largest immunoglobulin
  - It consists of 5-y shaped polypeptide chains<sup>bonds</sup> that held by J-chain
  - It is most efficient against the invading mechanisms
- first line defence

## \* IgD:

- It is composed of single y-shaped unit
- It function as B-cell receptor and present in a low concentration

## 33 \* IgE:

- It is a single x-shaped monomer
- It is tightly bind with mast cells
- These are elevated in individuals with allergies

## 7) Proto oncogenes →

\* The normal gene that present in cell is responsible for production of various factors and that proteins regulate the gene expression is known as protooncogenes

### → Mechanisms

The protooncogenes formation ~~under~~ occurs under 4 mechanisms and they are →

- i) point mutation
- ii) Chromosomal translocation
- iii) Gene amplification
- iv) Promoter insertion

# Protooncogenes

- i) point mutation
- ii) chromosomal translocation
- iii) Gene amplification

mutation

## Oncogenes

mutated proteins

- ↑ cell growth
- ↑ cell division

Cancer

-3-

need to add  
few points

Eg: protooncogenes

# 8) Collagen →

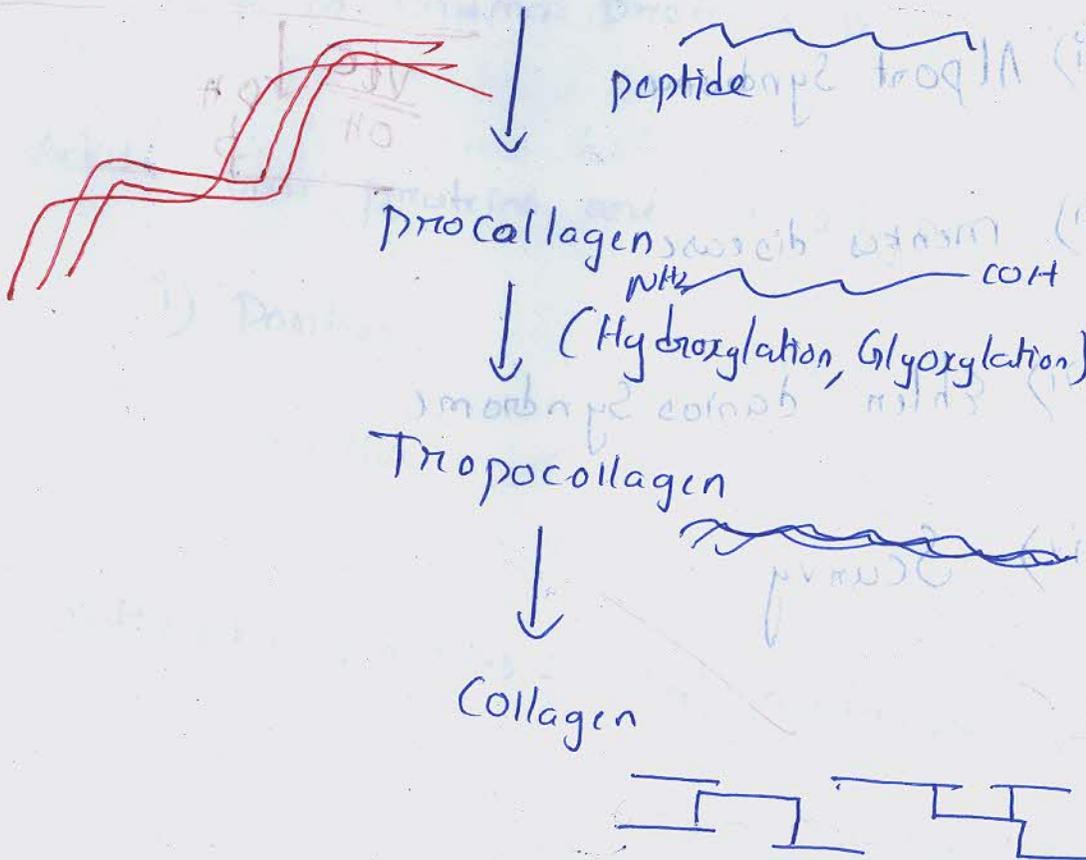
\* collagen is a type of structural fibrous proteins

\* Collagen synthesis occurs at PPTC

triple helical structure

Prepeptide

full form



\* It is most abundant protein found in animals

\* It is a triple helical structure

Stabilised by covalent linking

\* **Vitamin C** - Causes hydroxylation of proline.

\* Collagen fibres are of 19 types

+) proline & Glycine converted to hydroxy proline

→ Disorders

for

proline

i) Alport Syndrome



ii) Menkes disease

iii) Ehler Danlos Syndrome

iv) Scurvy

2

It is most abundant protein found in

## 9) Acute phase Proteins → non-specific response ⇐

- It refers to non specific response to stimulus of infection, injury.
- It is associated with characteristic pattern of change in plasma proteins.
- Acute phase proteins are of 2 types

i) Positive

ii) negative

→ Positive phase proteins →  $\alpha_1$ -antitrypsin  
 $\alpha_2$ -macroglobulin  
Ceruloplasmin  
C-reactive protein

→ Negative phase proteins → Albumin  
Transferrin

## ⇒ Ceruloplasmin

\* It is blue coloured copper containing  $\alpha_2$ -globulin

\* Its plasma concentration is around 30 mg/dl

\* It possesses oxidase activity

\* It is associated with Wilson disease

## ⇒ $\alpha_1$ -antitrypsin

\* It is a glycoprotein that contains 394 amino acids

\* Its molecular weight is around 54,000

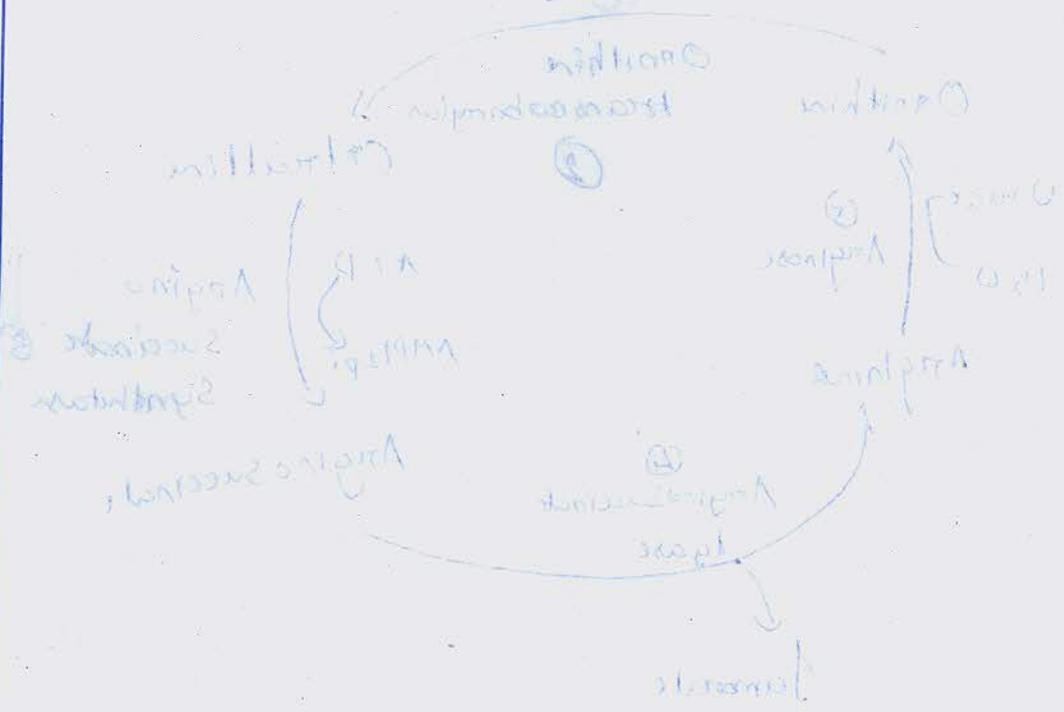
\* It is a serine protease inhibitor

⇒ C-reactive protein

Urea cycle

- \* It is a major component of acute phase protein
- > It is produced in liver
- > Increase in the C-reactive protein

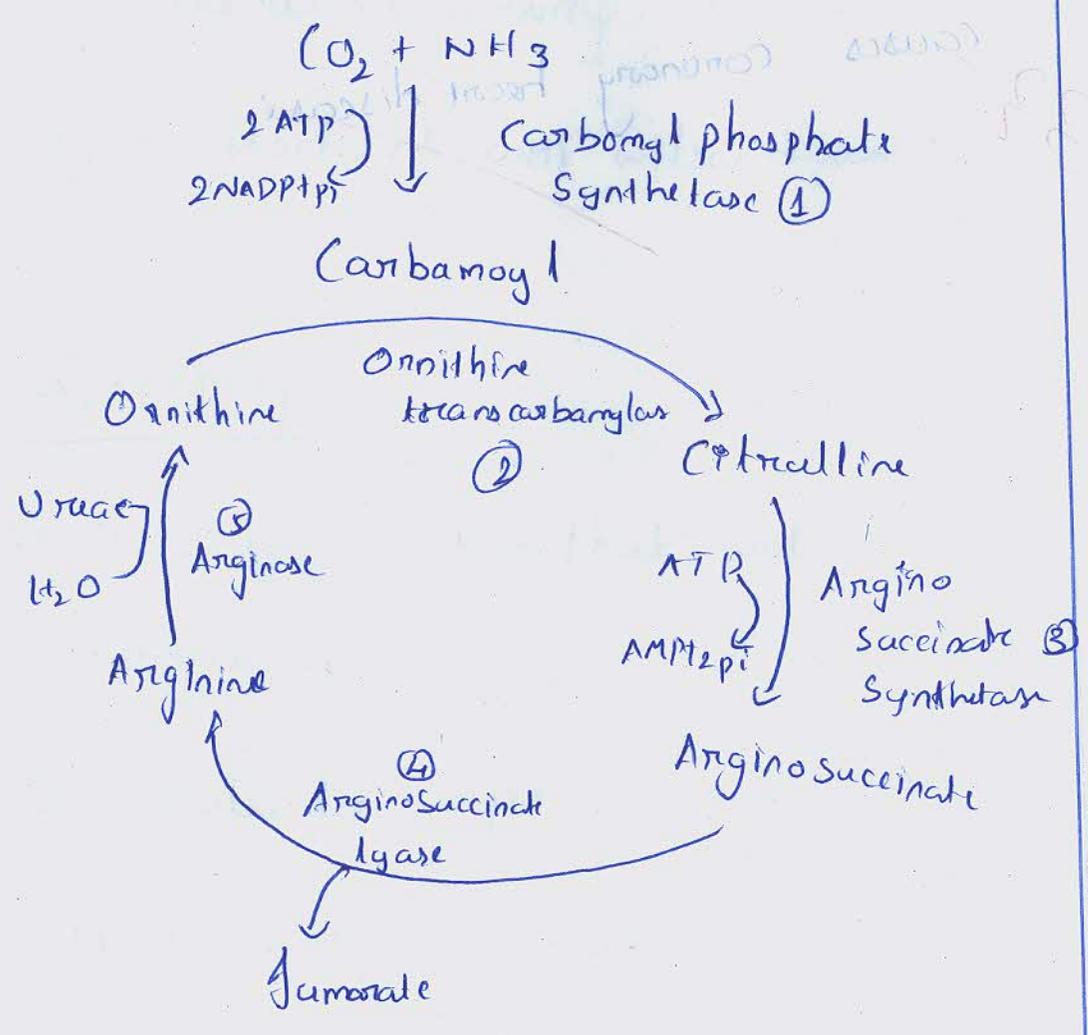
~~causes coronary heart disease~~



10) Urea Cycle → nitrogen excretion ⇌

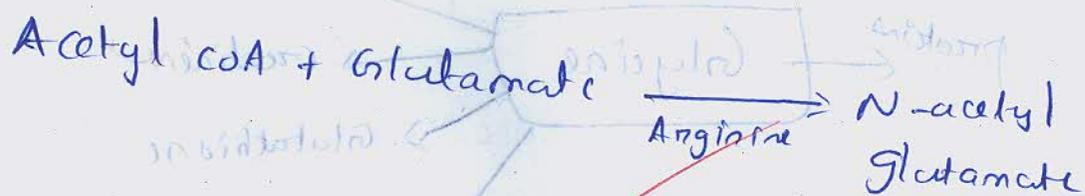
\* It is also known as ornithine cycle

\* 1<sup>st</sup> two reactions occur in mitochondria and remaining in the cytosol



⇒ Regulation

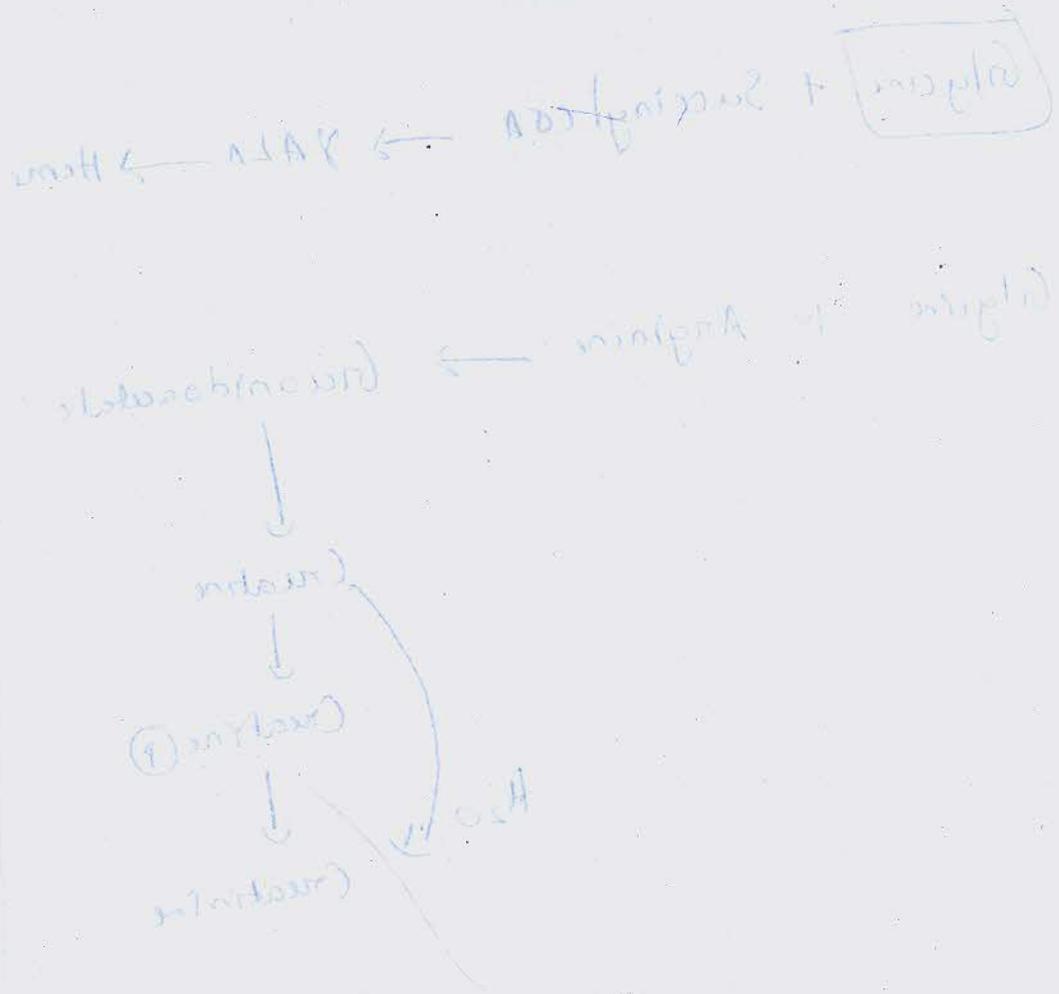
NAG - N acetyl Glutamate act as an allosteric inhibitor of (Carbamoyl) Synthetase



2 2

mention: Disorders

- Hyper ammoniaemia Type I
- Type II
- Enry
- CPS I
- Ornithine
- transcarbamylase

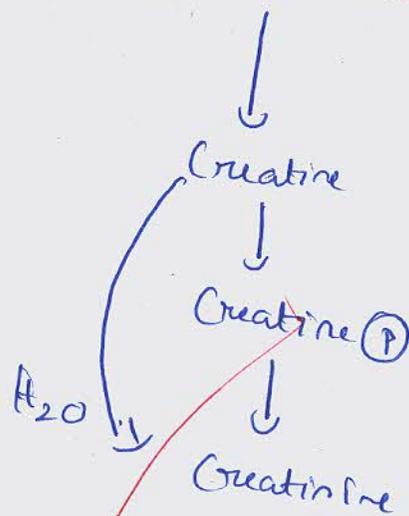
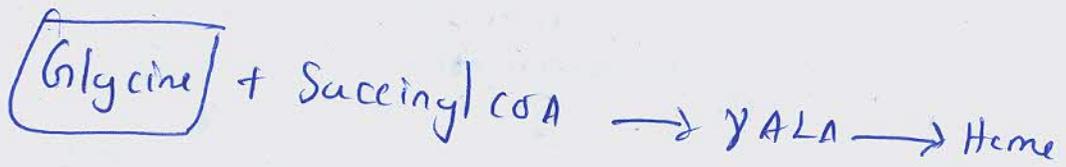


11)

a) Derivatives of glycine →



\* Glycine is a nonessential amino acid and it is optically inactive



5) SAM reactions →

1) Diagnosis →

on the basis of clinical characteristics  
we conclude that ~~best~~ best fit is

"Subtype from 2007"

ii) The affected metabolic pathway  
Purine nucleotide metabolism

→ Xanthine → Uric acid  
→ GMP → Guanosine

iii) Type & treatment

→ Type 2 → 2 types

i) Primary → IUPAC name

ii) Secondary → IUPAC name

12)

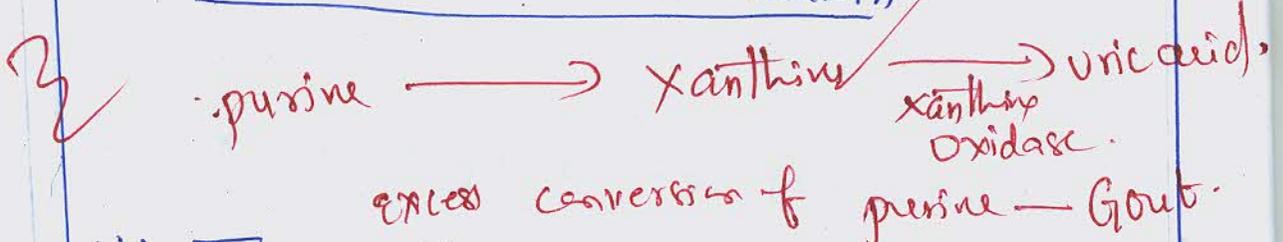
← another MA2 (d)

i) Diagnosis →

\* on the basis of clinical characteristics  
we conclude that ~~person~~ patient is  
suffering from 'Gout'

ii) The affected metabolic pathway

Purine nucleotide metabolism <sup>mention symptoms</sup> <sup>precipitating factor</sup>



iii) Types & treatment:

\* Types → 2 types

i) Primary → Inborn error

Enzymes:

ii) Secondary → Increase of uric acid

\* Treatment → Allopurinol  
Colchicine

iv) normal ranges →

males — 3-6 mg/dl  
female — 4-7 mg/dl

3

**IMPROVEMENT EXAMINATION – BIOCHEMISTRY**  
**PAPER -II**

Name:

Roll No:

10  
-----  
20  
cmdb

Answer all the MCQs

1x20=20 Marks

1. Which amino acid is required for both purine and pyrimidine synthesis (A) (B)  
A) Glycine      B) Aspartate      C) Alanine      D) Glutamate
2. In which aminoaciduria benedict's test is positive. (A) (C)  
A) phenyl ketonuria      B) Histidinemia  
C) Alkaptonuria      D) Nonketotic hyperglycinemia
3. Southern blotting is used to detect (A)  
A) DNA      B) RNA  
C) Proteins      D) Aminoacids
4. Frameshift mutation mutation results from (A) (B)  
A) Substitution of a single base      B) Deletion of a single base  
C) Addition of a codon      D) Deletion of a codon
5. Dystrophin is abnormal in: (D)  
A) Menke's kinky hair syndrome      B) Osteogenesis      C) Dermatosparaxis      D) Muscular dystrophy
6. The Amino acid which has maximum buffering capacity at physiology PH is: (C)  
A) Arginine      B) Alanine      C) Histidine      D) Glutamic acid
7. Fluidity of membranes depends on: (A)  
A) Nature of fatty acids      B) Concentration of Proteins  
C) Membrane pumps      D) Glycolation of proteins
8. Which of the following alpha globulins is a tumor marker? (A)  
A) Alpha fetoprotein      B) Thyroxin binding globulin  
C) Transcortin      D) High density lipoprotein
9. When pH falls by 1 unit, the hydrogen ion concentration: (A) (D)  
A) Decreases 10times      B) Increases two fold  
C) Changes by 7times      D) Increases 10times
10. Which Amino acid will give rise to an inhibitory neurotransmitter (B)  
A) Histidine      B) Glutamic acid      C) Ornithine      D) Tyrosine
11. The extacellular cation present in maximum concentration is: (C)  
A) Potassium      B) Magnesium      C) Sodium      D) Calcium
12. All the following Hormones have membrane receptors, except: (D)  
A) Insulin      B) Epinephrine      C) Glucagon      D) Thyroxine
13. Orotic aciduria is a feature of deficiency of all the following enzymes, except: (A) (B)  
A) OMP decarboxylase      B) Ornithine transcarbamoylase  
C) OPRTase      D) Aspartate transcarbamoylase
14. Zinc is present in all enzymes, except: (A) (B)  
A) Alkaline phosphatase      B) Amylase      C) Carbonic anhydrase      D) Carboxypeptidase
15. Hemopexin carries: (B)  
A) Free hemoglobin      B) Free heme      C) Free bilirubin      D) Free iron
16. Human insulin differs from bovine insulin in: (B) (D)  
A) Biological      B) Number of amino acids  
C) Position of disulfide bonds      D) Sequence of amino acids
17. Normal range of HCO<sub>3</sub> (C)  
A) 3.5-5.5 mmol/L      B) 136-145 mmol/L      C) 22-26 mmol/L      D) 96-106 mmol/L
18. Which is not true regarding the genetic code: (A) (B)  
A) Degenerate      B) Ambiguous      C) Nonoverlapping      D) Universal
19. All the following are channel formers, except: (C) (D)  
A) Gramicidin      B) Amelogenin      C) Valinomycin      D) Glutamate
20. Synthesis of recombinant DNA (r DNA) requires all the following, except: (C) (B)  
A) Restriction endonuclease      B) RNA primer  
C) Plasmid vector      D) DNA ligase